# Different Aggregation Behaviors of Tetra- (4-hydroxyphenyl) Porphyrin ( $\mathbf{T H P P H}_{2}$ ) in the Inner Core and on the Surface of CTAB Micelles 

Chen MA, Yun Hong ZHANG*, Chang Song FU, Qian Shu LI<br>School of Chemical Engineering and Materials Science, Beijing Institute of Technology, Beijing 100081


#### Abstract

The UV-Vis spectra of $\mathrm{THPPH}_{2}$ in CTAB micelles at pH 7.2 and pH 11.0 were analyzed to study the effect of micellar environments on the aggregation behaviors of this porphyrin.


Keywords: $\mathrm{THPPH}_{2}$, CTAB micelles, aggregation, UV-Vis spectra.

It is hard to study the effects of different microenvironments on porphyrin aggregation because of its complex substituted groups. A trans-membrane process was realized by controlling bulk pH values for an amphiphilic porphyrin from the inner core to the surface of CTAB micelles ${ }^{1}$. Thus it is posible to study the different aggregation actions of the porphyrin in the inner core and on the surface of micelle.

## Results and Discussion

The absorption spectra of $\mathrm{THPPH}_{2}$ at pH 7.2 have four Q bands at $516.8 \mathrm{~nm}, 555.7 \mathrm{~nm}$, 589.7 nm and 649 nm in Figure 1(A), with the maxma of the Soret band ( $\lambda_{\max }$ ) at 420.4 nm . When the porphyrin concentration is below $1.0 \times 10^{-5} \mathrm{~mol}^{2} \mathrm{~L}^{-1}$, the half band width of the Soret band is 14.3 nm , which is reduced to 12.3 nm above $1.0 \times 10^{-5} \mathrm{~mol} . \mathrm{L}^{-1}$. $\mathrm{THPPH}_{2}$ follows Beer's law in Figure 2, indicating that $\mathrm{THPPH}_{2}$ is solubilized in the inner core of CTAB micelles ${ }^{1}$., the absorbance $v s$ concentration plot does not follow Beer's law above $1.0 \times 10^{-5} \mathrm{~mol} . \mathrm{L}^{-1}$, indicating the occurrence of porphyrin aggregation ${ }^{2}$. The effect of aggregation on the absorption spectra often exhibits a change of the half band width ${ }^{3}$ and a shift of the Soret band ${ }^{4}$. There is no shift of the Soret band in Figure $\mathbf{1}(\mathrm{A})$, but the narrowing of the half band width suggests the formation of highly ordered porphyrin aggregate ${ }^{3}$. This aggregate may be some kind between H - and J-aggregate, in which the monomeric molecules arrange in a dimension that the angle between the transition moment and the line joining the molecular centers is 54 degrees in Figure 3 (A), resulting in the constant gap of the energy level of the Soret band ${ }^{3}$.

THPPH ${ }_{2}$ has two Q bands at 585 nm and 667 nm in Figure 1 (B), suggesting that the two N atoms on the pyrrole rings are also deprotonized, and the porphyrin molecules have transferred from the inner core to the outer surface of CTAB micelle ${ }^{1}$. $\lambda_{\text {max }}$ is 435.7 nm at lower porphyrin concentrations. At $1.28 \times 10^{-5} \mathrm{~mol} . \mathrm{L}^{-1}$, there is a plateau of the Soret band, and $\mathrm{A}_{422.7}$ is slightly higher than $\mathrm{A}_{435.7}$, suggesting the aggregate formation although it still seems to follow Beer's law (Figure 2). At higher concentrations, the Soret band has gradually shifted to 422.7 nm , and the plot does not
follow Beer's law, which consistent with an aggregation of $\mathrm{THPPH}_{2}$ on the outer surface of CTAB micelles. This aggregate might be considered a face-to-face H -aggregate ${ }^{3}$ from the narrowing and the blue shift of the Soret band in Figure 3 (B).

Figure 1. UV-Vis spectra of $\mathrm{THPPH}_{2}$ in CTAB solutions

(A) at $\mathrm{pH} 7.2,10^{-7} \mathrm{c}\left[\mathrm{THPPH}_{2}\right] /\left(\mathrm{mol}^{-1}\right): a .1 ; b .2 ; c .4 ; d .8 ; e .16 ; f .32 ;$ g. $48 ; h .64 ; i .80 ; j .96 ; k$. 104;l. 112; m. 120; n.128, and (B) at pH11.0, $10^{-7} \mathrm{c}\left[\mathrm{THPPH}_{2}\right] /\left(\mathrm{mol}^{-1}\right): a .1 ; b .2 ; c .4 ; d .8 ; e .16$; f 32;g.48; h. 64; i. 96; j. 128; k. 192; l. 256; m. 320.

Figure 2. Dependence of $\mathrm{THPPH}_{2}$ concentration at pH7.2 ( $\quad, \lambda \max =420.4 \mathrm{~nm})$ and $\mathrm{pH} 11.0\left(\nabla, \lambda_{1}=422.7 \mathrm{~nm} ; \square, \lambda_{2}=435.7 \mathrm{~nm}\right)$

Figure 3. Aggregation behaviors of $\mathrm{THPPH}_{2}$ in the inner core (A) and on the surface (B) of CTAB micelles



A
B

## References

1. Y.H. Zhang, L. Guo, Q.S. Li, Y.Q. Wang, Chem.J. ChineseUniversities, 1997, 18, 1703.
2. W.M. Clark, J. Biol. Chem., 1940, 135, 590.
3. D.C. Barber, R.A. Freitag-Beeston, D.G. Whitten, J. Phys. Chem., 1991, 95, 4074.
4. N.C. Maiti, M. Ravikanth, S. Mazumdar, J. Phys.Chem., 1995, 99, 17192.

Received 6 March 2000

